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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.008 Å R factor = 0.066 wR factor = 0.191 Data-to-parameter ratio = 7.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1,4-Dideoxy-2,3-O-isopropylidene-N-[3-(3-nitrobenzoylamino)benzoyl]-1,4-imino-D-talitol

The title compound, $C_{26}H_{29}N_3O_8$, has two molecules in the asymmetric unit. Molecules are linked into chains by N-H···O hydrogen bonds, and C-H···O interactions are observed between chains.

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organic papers

Comment

The title compound, $C_{26}H_{29}N_3O_8$, (I) (Fig. 1), is a derivative of 1,4-dideoxy-2,3-*O*-isopropylidene-1,4-imino-D-talitol (Fleet *et al.*, 1988). The 3-aminobenzoic acid group can be considered as a β -amino acid with a rigid conformation (Pohl *et al.*, 2001). Thus, the crystal structure of (I) provides information for conformational analysis of this new type of β -peptide mimetic.



Two molecules of (I) exist in the asymmetric unit (Fig. 1). Apart from the dimethyldioxolanyl groups, C5/C6/C7/O3/O4/ C11/C12 and C31/C32/C33/O13/O21/C37/C38, the molecules are related by pseudo-inversion symmetry. $N-H\cdotsO$ hydrogen bonds are formed between the peptide linkages (Table 1), generating $C_2^2(14)$ chains (Bernstein *et al.*, 1995) propagating along the *a* direction. Numerous $C-H\cdotsO$ interactions are observed between chains (Table 1).

Experimental

The title compound, (I), was prepared from 1,4-dideoxy-2,3-Oisopropylidene-1,4-imino-D-talitol in three steps, as follows. In the first step, 1,4-dideoxy-2,3-O-isopropylidene-1,4-imino-D-talitol (4.5 g, 18.5 mmol) was dissolved in a mixture of pyridine (10 ml) and CH₂Cl₂ (20 ml). A solution of 3-nitrobenzoyl chloride (3.7 g, 20 mmol) in CH₂Cl₂ (10 ml) was added slowly to the reaction mixture at 273-278 K (ice bath). After stirring for 2 h, the solvents were removed under reduced pressure and the residue was then dissolved in ethyl acetate (20 ml). The solution was washed with a saturated aqueous solution of NaHCO₃ (20 ml). The organic layer was separated and the aqueous layer was extracted with ethyl acetate (2 \times 20 ml). The combined organic layers were dried over MgSO₄, then concentrated in vacuo. The residue was purified by column chromatography on silica gel (petroleum ether/ethyl acetate, 60:40) to give N-(3-nitrobenzoyl)-1,4-dideoxy-2,3:5,6-di-O-isopropylidene-1,4-imino-D-talitol as a colourless gum (6.8 g, yield 94%).

© 2006 International Union of Crystallography All rights reserved In the second step, a solution of N-(3-nitrobenzoyl)-1,4-dideoxy-2,3:5,6-di-O-isopropylidene-1,4-imino-D-talitol (1.36 g, 3.5 mmol) in ethanol (20 ml) was stirred under an H₂ atmosphere in the presence of 10% Pd on charcoal (150 mg) at room temperature for 5 h. The reaction mixture was then filtered through celites to remove the catalysts. The filtrate was concentrated under reduced pressure to give the crude aniline, which was directly dissolved in a solution of pyridine (3.5 ml) and CH₂Cl₂ (20 ml) and cooled with an ice–water bath.

In the third step, a solution of 3-nitrobenzoyl chloride (1.0 g, 5.4 mmol) in CH₂Cl₂ (10 ml) was added dropwise under an Ar atmosphere. After stirring for 2 h, the solvents were removed under vacuum and the residue was dissolved in ethyl acetate (20 ml). The resulting solution was washed with a saturated aqueous solution of NaHCO₃ (20 ml). The organic layer was separated and the aqueous layer was extracted with ethyl acetate (2 × 20 ml). The organic layers were combined, dried (Na₂SO₄) and evaporated under reduced pressure to give the crude product, which was purified by column chromatography on silica gel (petroleum ether/ethyl acetate, 1:1) to give the title compound (1.5 g, yield 85%). Suitable single crystals were obtained by slow evaporation of a solution in ethyl acetate (m.p. 456–457 K).

Z = 4

 $D_r = 1.294 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Block, colourless

 $0.24 \times 0.20 \times 0.14 \text{ mm}$

21856 measured reflections

5372 independent reflections

3777 reflections with $I > 2\sigma(I)$

 $\mu = 0.10 \text{ mm}^{-1}$

T = 294 (2) K

 $R_{\rm int} = 0.043$

 $\theta_{\rm max} = 26.0^{\circ}$

Crystal data

 $\begin{array}{l} C_{26}H_{29}N_{3}O_{8}\\ M_{r}=511.52\\ \text{Monoclinic, }P2_{1}\\ a=12.5459\ (13)\ \text{\AA}\\ b=14.5417\ (14)\ \text{\AA}\\ c=14.6939\ (15)\ \text{\AA}\\ \beta=101.563\ (4)^{\circ}\\ V=2626.3\ (5)\ \text{\AA}^{3} \end{array}$

Data collection

Rigaku Saturn CCD area-detector diffractometer ω scans Absorption correction: multi-scan Jacobson (1998) $T_{\rm min} = 0.977, T_{\rm max} = 0.987$

Refinement

$w = 1/[\sigma^2(F_0^2) + (0.1155P)^2]$
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.41 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.19 \text{ e } \text{\AA}^{-3}$
Extinction correction: SHELXL97
(Sheldrick, 1997)
Extinction coefficient: 0.019 (3)

Table 1	
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		0	
Hydrogen-bond	geometry	(A,	°)

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N2-H2A\cdotsO14^{i}$	0.87 (6)	2.13 (6)	2.981 (6)	165 (5)
$N5-H5A\cdots O5^{ii}$	1.10 (6)	1.90 (6)	2.982 (6)	166 (5)
$C24-H24\cdots O13^{iii}$	0.93	2.59	3.489 (7)	162
$C25-H25\cdots O8^{iv}$	0.93	2.52	3.418 (7)	163
$C51 - H51 \cdots O17^{v}$	0.93	2.56	3.473 (7)	166
$C52-H52\cdots O5^{ii}$	0.93	2.56	3.370 (7)	146

Symmetry codes: (i) -x + 1, $y + \frac{1}{2}$, -z + 1; (ii) -x, $y - \frac{1}{2}$, -z + 1; (iii) x + 1, y, z + 1; (iv) -x + 2, $y + \frac{1}{2}$, -z + 2; (v) -x - 1, $y - \frac{1}{2}$, -z.





A view of the asymmetric unit of (I), showing displacement ellipsoids at the 30% probabilty level. H atoms bound to C atoms have been omitted.

H atoms were placed in calculated positions and refined using a riding model, with C–H = 0.93 Å for Csp^2 , 0.98 Å for CH, 0.97 Å for CH₂ and 0.96 Å for CH₃. $U_{iso}(H) = 1.2U_{eq}(C)$, except for the methyl groups, for which $U_{iso}(H) = 1.5U_{eq}(C)$. The H atoms of the N–H groups were located in difference Fourier maps and refined with isotropic displacement parameters. The displacement parameters of atoms C37 and C38 were restrained to approximate isotropic behaviour. In the absence of significant anomalous scattering effects, Friedel pairs were merged as equivalent data for the final cycles of the known configuration of the starting material.

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997) and *DIRDIF99* (Beurskens *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *CrystalStructure* (Rigaku/MSC, 2005); software used to prepare material for publication: *CrystalStructure*.

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